

# Why the maximum quantum efficiency(QE) of a single junction type Solar cell is about 35 % ?

W. Shockley and H. Queisser, " J. Applied Physics, 32(1961)510

- A photon with an energy less than the band gap( $E_g$ ) cannot be converted to the electron energy. QE approaches zero as  $E_g$  gets larger and larger.
- A single junction type semiconductor with a very small band-gap has a very small band-bending, resulting in a very small barrier electric field for separating photo electron and hole pairs. QE approaches zero as  $E_g$  gets smaller and smaller.
- Light with a wave length of  $0.4 \mu\text{m}$  has a photon energy of  $E = 1.24/0.4 = 3.1 \text{ eV}$ . In a silicon-based solar cell with the band gap of  $1.1 \text{ eV}$ , carriers are needed in order to maintain the hot electron energy of  $3.1 \text{ eV}$  from being reduced down to  $1.1 \text{ eV}$  by collisions with other free electrons and orbit electrons in the bulk silicon crystal, before reaching the external output load. The maximum value of QE would be less than  $1.1/3.1 = 35.5\%$  for a silicon-based type solar cell.
- Short-wave-length high-energy photons have a very short light penetration depth in the silicon crystal. In case of a floating-surface N+PP+ single junction type solar cell, there is no surface electric field in the surface N+ diffusion region. And the case becomes worse since the photo electron and hole pairs, which have been generated at the surface vicinity, stay together at the silicon surface. Soon or later eventually, all of the photo electron and hole pairs are recombined and wasted into heat. In case of the single junction type silicon-based solar cell, QE becomes very small.

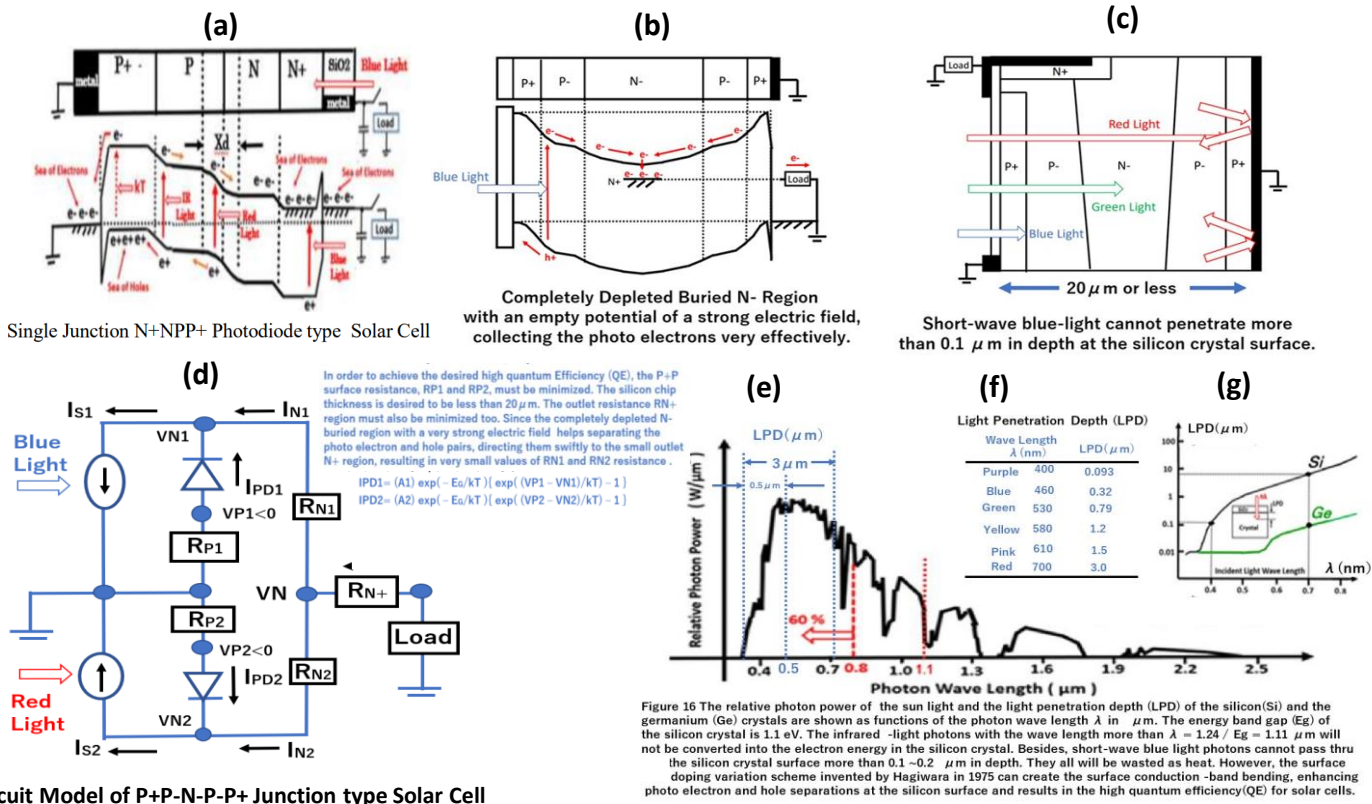


Figure 16 The relative photon power of the sun light and the light penetration depth (LPD) of the silicon(Si) and the germanium(Ge) crystals are shown as functions of the photon wave length  $\lambda$  in  $\mu\text{m}$ . The energy band gap ( $E_g$ ) of the silicon crystal is  $1.1 \text{ eV}$ . The infrared -light photons with the wave length more than  $\lambda = 1.24 / E_g = 1.11 \mu\text{m}$  will not be converted into the electron energy in the silicon crystal. Besides, short-wave blue light photons cannot pass thru the silicon crystal surface more than  $0.1 \sim 0.2 \mu\text{m}$  in depth. They all will be wasted as heat. However, the surface doping variation scheme invented by Hagiwara in 1975 can create the surface conduction -band bending, enhancing photo electron and hole separations at the silicon surface and results in the high quantum efficiency(QE) for solar cells.

**Blue light has a very short Light Penetration Depth (LPD) of less than  $0.05 \mu\text{m}$ .**